



Evaluation of Indirect Liquid Cooling for Electric Vehicle Battery Thermal Management System

by

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(Supervised by Professor Youn Cheol Park)

DISSERTATION

Submitted in partial fulfillment of the requirements for the degree of Master's in Mechanical Engineering

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In the Name of Allāh, the Most Gracious, the Most Merciful

بالله وَمَا إلا

My success is only by Allah

رد زدني عِلْمًا

My Lord! Increase me in knowledge.



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Nomenclature

А	Ampere
Ah	Ampere hour
BEV	Battery electric vehicle
BTMS	Battery thermal management system
C_f	Smooth pipe friction factor
СМН	Cubic meters per hour
C _p	Specific heat
d	Pipe diameter
EV.	Electric vehicle
HEV	Hybrid electric vehicle
HTF	Heat transfer fluids
К	Thermal conductivity
k	Thermal Conductivity
kW	Kilowatt
kWh	Kilowatt per hours
Li-B	Lithium-ion battery
Li-ion	Lithium-ion
LPM	Liter per minute
Ni-Cd	Nickle-cadmium
Ni-MH	Nickle-metal hybrid



Nu Nusselt number

- PCM Phase change materials
- PHEV Plug-in hybrid electric vehicle
- Pr Prandtl number
- TR. Thermal Runaway
- *U*_{eff} Effective velocity outside boundary layer
- ρ Density



Abstract

Green-powered electric vehicles are reliable substitutes to lessen greenhouse gas emissions and reliance on fossil fuels. Electric vehicles are now extensively used worldwide, and this will be high in the future. The battery pack is the key feature of an electric vehicle. The battery pack's performance and life are incredibly temperature-sensitive, a problem related to battery thermal management. So, it is mandatory to keep the optimal temperature range for smooth performance and safety. This study evaluates the coolant cooling system for the thermal management of 45 kWh and 64 kWh battery packs, respectively. A chiller and radiator are employed to cool the operating fluid, combining water and ethylene glycol with the 50:50 ratio. A thermal-fluid simulation executed using a standard commercial tool known as GT Suite, and the battery pack temperature experiment using a vehicle was conducted to validate the simulation results. The results reveal that the proposed systems' performance typically improved by improving the flow rate and the chiller power. It depicts that the cooling rate is maximum at 40 °C initial battery temperature and an ambient temperature of 25 °C with 3 kW chiller capacity and flow rate of 20 liters per minute (LPM) in the chiller cycle mode. While in the radiator cycle, the highest cooling rate is at an ambient temperature of 25 °C, an initial battery temperature of 40 °C with an airflow rate of 3600 cubic meters per hour (CMH), and a 30 LPM of coolant flow rate in both systems, respectively. Besides, a thermal management system of a 45 kWh battery pack was found better than a 64 kWh battery pack system



in terms of cooling performance. In the future, the proposed BTMS could be a realistic solution and help develop the electric vehicles' thermal system.



Chapter 1 Introduction

1.1 Overview

In the 21st century, with growing concerns about the scarcity of fossil fuel, the price of crude oil, and technological development, electric vehicles have acquired a greater interest in transport mode. Many electric vehicles, including pure electric vehicles (EVs), hybrid electric vehicles (HEVs), and plug-in hybrid electric vehicles (PHEVs), have been produced in recent years. Toyota and Honda have introduced the Prius, Insight, and Civic electric vehicles since 1997. The consumers of these vehicles are praised tremendously. In 2008, Tesla launched the first-ever lithium-ion battery electric vehicle, namely Tesla Roadster, for over 200 mileage. Nissan Leaf, KIA NIRO, and Hyundai Kona gained tremendous attention in recent years. According to the International Energy Agency, electric cars (BEVs, PHEVs) deliveries topped 2.1 million worldwide to boost the total stock to 7.2 million electric vehicles in 2019 [1].

The global challenge of electricity scarcity and environmental emissions has prompted the increasing popularity of electric vehicles to take the top stage in the future. In the transport market, electric vehicles are expected to increase 22% by 2030 due to emissions from EVs that have up to 43% lower emissions than diesel vehicles and have lower lifetime climate impact than those with IC engines. Anderson et al. [2] indicated that the EVs' usage could lead to a remarkable reduction of greenhouse gas emissions by approximately 40% if the energy supplies by renewable resources.



Electric vehicles are becoming increasingly prevalent worldwide to solve the deterioration of environmental and energy problems. The effectiveness of electric vehicles relies primarily on their storage system for electricity. The significant areas of focus for electric vehicles are to find a reliable system of energy storage that can accommodate high mileage, rapid charging, and practical driving [3]. Currently, lithium-ion batteries have the best characteristics to satisfy the wide variety of specifications unique to automotive applications low cost, no pollution, self-discharge rate, good power capabilities, light-weight, no memory effect, higher energy density, and long-life cycle [4]. The flat-plate batteries are often preferred because of their higher power density.

1.2 Adverse Effect of Temperature on Lithium-ion Battery

Compared to other rechargeable batteries, such as lead-acid, nickel-cadmium (Ni-Cd), nickel-metal hybrid (Ni-MH) batteries, rechargeable lithium-ion batteries are considered the most suitable energy storage option for EVs. However, Li-ion batteries may be problematic if they are not appropriately treated. The battery's performance and state of health (SOH) is firmly dependent on the working ambient temperature.

A large quantity of heat is generated inside the pack, particularly during the fast charging and discharging process because of several chemical reactions and phenomena of electrochemical transport, which will increase the battery temperature.



The generated heat is maximum as hundreds of cells are serially or parallel connected and battery irregularity and thermal conditions causing the temperature difference between them. The exothermic existence of all of these reactions shows that a battery pack's performance is affected by temperature. The heat is contained within the battery pack if it is not adequately dissipated. The author of Ref. [5] observed that the voltage, electricity, efficiency, and life cycle of a cell are profoundly affected by its operating temperature.

Temperature uniformity significantly affects the health of the battery pack in the longterm. For li-ion batteries, there are two essential items:

First, the working temperature goes above the appropriate temperature limit. Second, the uniformity of the low-temperature contributes to the loss of battery life.

1.2.1 Low Temperature

Operating a lithium-ion battery below the optimal temperature range can cause battery performance reduction, leading to limits in its use in a cold environment. There are many issues with low ambient temperature, such as high resistance to receiving charging, battery life span, and round-trip efficiency. The charger cannot charge the battery pack fast if the battery is working below 5 °C. If the battery is running below 0 °C, it will lose charge, power, acceleration, and driving range. It identified that if the battery is working at a lower ambient temperature over the longrun, the battery dendrites the anode lithium, causing a decline in battery safety. Smart



et al. [6] showed that it is tough to charge the battery at its maximum capacity because lithium plating can occur at high charging levels at room temperature. The author already discussed the impact of low-temperatures on battery pack health in previous studies [7,8].

1.2.2 High Temperature

In a vehicle, an entire battery pack typically comprises multiple single cells attached serially or in parallel. Many battery cells can produce a significant amount of heat during the charge and discharge process, allowing temperatures to increase [9]. There are various effects exerted on the battery's performance at the temperature above the optimal, such as self-discharge, power fade, and these could cause the accessible energy to experience enormous losses [10]. It may be even worse when hundreds of cells are bound in series or parallel.

Battery running above the optimal ambient temperature can induce volume swelling, and non-uniform distribution of temperatures affects the batteries thermally and results in various forms of a node. During spirited driving and without conditioning, battery life can be decreased by 2/3 in hot weather conditions [11]. At an operational temperature between 30 °C to 40 °C, the lithium-ion battery life reduces by 60 days per degree of temperature rise [12].

The author of Ref. [13] experimented to determine the influence of temperature and aging time on the Li-ion cells. The results revealed that above the short course of 20



weeks at 55 °C, a power decline of 55% was obtained. Several complex reactions occur as batteries are charged/discharged. Electrochemical reactions affect the heat generation rate, and the speed of the electrochemical reactions is further increased by higher temperatures [14]. That is one of the main reasons for batteries' thermal behavior, related to these reactions [15].

For EVs, safety is a concern and a serious problem. In addition to the low high temperature's performance, each module's cell is susceptible to short circuits overheating and extreme ambient temperature. With the advent of fire and explosion, the thermal runway (TR) may occur when the self-healing process is out of control. Besides, the variation in temperature between each cell will lead to a low rate of variable power, decreasing the battery pack's life cycle. In one study [16], the author identified that charging performance and battery life deteriorates the most quickly due to heat at temperatures above 50 °C. As the battery temperature increases constantly, heat cannot be extracted in time after the temperature exceeds 80°C thermal runway will occur [17].

Thermal runaway is mostly caused by short-circuiting [18] and overheating [19]. Extreme exothermic reactions occur one after another as the thermal runaway occurs: layer decomposition of solid electrolyte interphase (SEI), adverse electrodeelectrolyte reaction, decomposition of electrolyte, and positive electrode-electrolyte reaction [20]. E.g., LFP, decomposed SEI layer at 100°C, melting separator, and



shrinking at 130°C, allowing the electrodes to have short circuits. With the increasing temperature, the widely used positive electrode material would continue to deteriorate and begin to produce oxygen. The released oxygen and heat during the reaction process cause the fire and explosion for batteries and the thermal runaway occurs at temperatures above 150°C [21]. Furthermore, a significant quantity of poisonous gas can endanger personal protection with a sudden release of a large amount of energy.

Three factors [22] typically cause a single battery thermal runaway to occur: one is the loss in internal thermal resistance, such as inflammation gas in the battery core, which will significantly increase thermal resistance. (2) The fault of heat generation, such as a short-circuit that results in an uncontrollable generation of heat. (3) The outer convective cooling flaw, such as when in an adiabatic condition, battery heat could not be removed to the outside.



Figure 1. Process of thermal runaway [23].

The characteristics of the particular battery need to be understood to maximize their

performance.



1.2.3 Advised Temperature Range

Batteries are susceptible to several problems, for example, short circuits and high ambient temperatures leading to battery overheating and failure. Thus, it's vital to design an actual thermal system to regulate a specified range of battery temperature. To obtain the optimal temperature range, many researchers have advised the optimum temperature range. For example, the author of Ref [24] presents that commercial li-ion battery packs achieve optimum performance near room temperature. Pesaran et al. [25] advised the narrow temperature range of 25 ~ 40 $^\circ$ C for lithium-ion batteries, and the desired cell to cell or module to module temperature difference is 5 °C. Authors of [26] present an increase in the temperature difference over 5 °C between the battery pack and a 10% reduction in power. He at al. [27] suggested that a typical temperature range is between 20 °C and 40 °C, and the longrange for lithium-ion batteries is between -10 °C and 50 °C for tolerable operation [28]. Finally, to sum up, at a temperature between 25 °C and 40 °C, lithium-ion batteries work best. In a battery or from module to module, less than 5 °C is the optimal temperature distribution.

1.3 Battery Thermal Management System

They expose high or low temperature, voltages, excessive fast-tracked battery degradation, and battery failure. According to the working environment and conditions, a thermal management system must ensure reliability, performance,



battery life, range, and safety to prevent runaway. A thermal management system of the battery pack is also required to provide a reasonable temperature spread.

There are two approaches to decrease the chance of thermal runaway and to boost thermal stability; one is affected by the electrode characteristics, such as electrode material [29], electrolyte [30], and separator [31]. The properties of battery thermal runaway processes vary with the battery material variation [32], and this is far away. Another solution is to apply a thermal system, called a battery thermal management system, to the batteries for a suitable operating environment. Thus, developing an effective BTMS for packs is the only way to avoid thermal runaway in batteries.

BTMS has two main functions: To prolong battery life and improve electrical performance by keeping batteries within an optimal temperature range, having improved safety to prevent thermal runaway.

Moreover, two things are expected in BTMS: Battery temperature should be within the optimal range and to keep the non-uniform temperature within the required range between the cells. BTMS has to be combined with the battery pack.

Furthermore, there are few things to consider before designing the battery pack's thermal management system; What is the proposed system's cost? What is the optimal temperature range? Which kind of heat transfer pattern is appropriate?



Which method of cooling is suitable? Direct or Indirect? How much heat removes from the pack?

The factor considered in the analysis includes ease of operation, capital cost, energy efficiency, reliability, and maintenance requirement.

1.3.1 BTMS Selection Based on Medium



Figure 2. Available battery thermal management system according to the medium.

As discussed above, the battery thermal management system keeps narrow temperature bandwidth and prevents thermal runaway potential in extreme temperature conditions.



Several methods to quickly extract heat from a battery pack have been evaluated by numerous research groups worldwide in recent years. As seen in Figure 2, they can categorize into several forms. A battery thermal management system can be distinguished by the operating fluid used in the cooling loop.

Some popular BTMS systems are:

- Air Cooled
- Liquid Cooled
- Phase change materials (PCM)
- Hybrid (Any other)

The air-cooled system can be a natural or forced air cooling system depending on the requirement. The liquid cooling system is divided based on the operating mode; there are two modes: Direct mode and Indirect mode. In the first one, the liquid directly contacts the battery to remove the heat. One example of this system is jacket cooling. This system is usually employed for cylindrical batteries.

On the other hand, in the indirect cooling method, a mini channel, discrete tube, or cold plate is considered to remove the battery's heat by contact surface strategy. Another one is phase change materials based BTMS. As shown by name phase change materials, the PCM cooling method is used to remove the heat. Materials evaporate by the evaporation method and then condensed in the condenser. By evaporation and condensation processes system pulled the heat from the pack.



Low thermal conductivity and low specific air capacity make air cooling limited, and the ability to reduce and maintain the temperature is low. PCM is massive, costly, and not eco-friendly. The heat collected by PCM will also need to be extracted by other cooling techniques since a given mass of overall heat capacity is limited [33]. Besides, phase change material's thermal conductivity is low, resulting in poor performance of the system's heat conduction. Heat pipe transfers heat through latent heat as it is in its initial stage of studies.

On the other hand, the higher specific heat power and higher heat exchange efficiency of liquid coolant make it simple than air to satisfy the battery's thermal requirements. The liquid cooling has a more robust heat exchanger capacity than air-cooled BTMS. In addition, because of its higher heat transfer coefficient, a liquid cooling system is more potent in cooling, and less pumping power is needed than the air system. Besides, a suitable battery management system's choice relies on many aspects, such as working efficiency, installation costs, and volumetric constraints. The liquid cooling system has a compact structure that allows taking less volume and narrow space. Due to its high efficiency, simple maintenance, and technological sophistication, liquid cooling is commonly employed.

The author of Ref. [34] studies that a liquid-based thermal management system can minimize power consumption and have a better cooling impact relative to the aircooling approach. Moreover, it has been employed in commercial electric vehicles, such as BM i3 and i8, Chevrolet Volt, Tesla Model S, Kia Niro, and Hyundai Kona.



Liquid cooling maybe 3500 times more efficient or converse up to 40% of parasitic energy than air cooling [35]. Due to the advantages mentioned earlier, the liquid cooling method is considered in this research to cool the battery.

1.3.2 Indirect Cooling Method

The battery's direct contact with the heat transfer fluid might not be feasible in battery packs, considering the high thermal performance of direct-contact mode [36]. Furthermore, it's better to consider an indirect liquid-based system. The lower viscosity of a heat transfer fluid (e.g., ethylene glycol/water) than a dielectric liquid (e.g., mineral oil), resulting in a much higher flow rate for a defined pumping capacity [37]. Thus, passing the fluid through a channel, a metal plate with a cold plate is commonly used in commercial EVs. The performance of an indirect liquid cooling battery pack was evaluated by Chacko et al. [38]. They concluded that one of the most promising ways of achieving battery thermal management would be successful indirect liquid cooling/heating.

1.3.3 Cold Plate

A way of cooling a battery pack is to circulate a liquid coolant via a jacket or separate tubes around it or placing it directly on a liquid-cooled plate. While designing the indirect cooling system, few concerns need to be counted. The cold plate is suitable for prismatic and pouch cells. Jacket cooling is ideal for cylindrical cells. Pouch cells



are employed in this study and based on the cell type and volume, and a cold plate is suitable as a way of cooling. The cooling performance of a cold plate depends on many things such as; the diameter of the inlet and outlet of channels, the number of channels, channel shape, the position of the cooling plate, the geometry of the flow path, and flow direction.

Jin et al. [37] presented an ultra-thin cold plate for EV batteries' thermal management, useful in the cooling performance. To enhance the pack's thermal uniformity, Zhao [39] provides two significant promises: shortening the flow paths by increasing the flow channels and increasing the contact area between the battery and the cooling channels. The flow influences the cooling effect for large battery sizes because of the distinct heat generation rate distribution.

The channel's physical dimension, such as its length, width, or path, affects cooling plates' impact. The author of Ref. [40] explored several optimization methods for liquid cooling plate channel quantity, mass flow rate, flow direction, and input size to monitor the difference in maximum temperature and battery temperature.

The structure of liquid cooling makes it simpler, supported by Pesaran et al. [41]. Huo et al. [42] found that the direction of liquid flow, the flow rate, and the number of channels significantly influence the straight cooling channel's cooling effect. They gave recommendations on the number of channels needed for the flow rate. The impact of different serpentine structures on temperature uniformity and the cooling plate's average temperature was studied by [43].



The cold plates' cooling performance with serpentine channels was examined by Panchal [44]. The coolant flow paths were identical to the Chevrolet Volt cold plate configuration. The findings revealed that higher discharge rates from 1 coulomb to 4 coulomb and conditions from 5 °C to 35 °C resulted in higher battery temperature contours.

1.4 Objective/Aim

This study aims to evaluate the liquid cooling performance for the thermal management of the battery pack. The main goal is to develop a cooling system that can achieve high efficiency, ease of maintenance, low power consumption, low cost, and reliability by using the indirect liquid cooling technique. In other words, the main focus is to find the optimal values to design the BTMS. A chiller and a radiator with the flow rate variations are used to remove the system's heat. The findings indicate the rise in cooling rate with the increase in flow rate and the chiller capacity and radiator airflow rate. A BTMS that can track, control, maintain, and ensure the battery pack safety is considered. As mentioned above, the literature pays more attention to study the thermal management of the battery pack.

1.5 Thesis Structure

The rest of this dissertation is structured as follows: chapter 2 elaborates the system description, chapter 3 presents the test conditions, chapter 4 discusses the result and



discussion, chapter 5 is about comparison and verification of the obtained results, and finally, chapter 6 concludes the study with the benefits, drawbacks, future work and suggestions.



Chapter 2 Modeling, Simulation, and System Description

2.1 Conservation Equations

The continuity equation is:

$$\frac{dm}{dt} = \sum_{boundaries} \dot{m} \tag{1}$$

Where \dot{m} is the boundary mass flux into volume, & \dot{m} = pAu, and m = mass of the volume. The energy conservation equation for the explicit solver is:

$$\frac{d(me)}{dt} = -P\frac{dV}{dt} + \sum_{boundaries} (\dot{m}H) - hA_s \left(T_{fluid} - T_{wall}\right)$$
(2)

While the enthalpy conservation equation for the implicit solver is:

$$\frac{d(pHV)}{dt} = \sum_{boundaries} (\dot{mH}) + V \frac{dp}{dt} - hA_s \left(T_{fluid} - T_{wall}\right)$$
(3)

Where e is the total specific internal energy (internal energy plus kinetic energy per unit mass), *P* is the pressure, *V* is the volume, *H* is the total specific enthalpy, A_s is the heat transfer surface area, T_{fluid} is the temperature of the fluid, and T_{wall} is the wall temperature. The momentum conversation equation is:

$$\frac{d\dot{m}}{dt} = \frac{dpA + \sum_{boundaries} (mu) - 4C_f \frac{pu|u|}{2} \frac{dxA}{D} - K_p \left(\frac{1}{2} pu|u|\right)A}{dx}$$
(4)

Where *u* is the velocity at the boundary, C_f is Fanning friction factor, d_x is the length of the mass element in the flow direction, and d_p is the pressure differential acting across d_x .



2.2 Required Operating Mode of Thermal System

A battery thermal management system is necessary in order to maintain the battery pack within the optimal temperature range. The next question is about the mode; which mode is suitable according to the battery pack's initial temperature. There are many modes to keep the battery temperature within the optimum range, as shown in Figure 3. The selection of operating mode depends on the required temperature. Automatic control techniques are used to change the operating mode as per the instructions.

A chiller mode is required to down the temperature if the battery's initial temperature is above 32 °C. Whereas, if the battery's initial temperature is above 24 °C, a radiator mode is suitable. Besides, if the temperature is in the range of 8 ~ 23 °C, a bypass mode will be applied. Finally, if the temperature is below the 6 °C, a heater mode must maintain the battery pack's optimal temperature range.





Figure 3. BTMS operating mode required related to battery temperature flow chart.



2.2.1 Chiller Mode

The liquid coolant flow is operated by a three-way expansion valve, based on the cooling load and the exterior environment. The chiller mode cycle is straightforward; it consists of a chiller, pump, control valve, and a battery pack. Figure 4 illustrates the chiller cycle used in this study. A mixture of water and ethylene glycol is used as a coolant to remove the battery pack's heat. In the chiller mode, where the battery requires high cooling performance or where the ambient temperature is higher than the liquid coolant temperature, the coolant flows to the chiller to exchange heat. The chiller with three different capacities, 1, 2, and 3 kW, is used in the study. Simultaneously, the coolant flow rate was 5, 10, 20, and 30 LPM.



Figure 4. Cooling system schematic diagram using chiller.



2.2.2 Radiator Mode

In the radiator mode, a radiator is considered to work as a heat exchanger to remove heat to the environment. The liquid coolant flows to the radiator cooled by the outside air, while the battery requires low cooling performance, and the ambient temperature is lower than the liquid coolant temperature. The chiller is bypassed in this cycle because of less need for the temperature difference. A radiator mode cycle is illustrated in Figure 5. This cycle consists of a battery pack, a pump, a control valve, a radiator, and valves to bypass the coolant. It depends on the ambient temperature of the active region. A fan is also employed to circulate the air through the radiator to enhance the heat transfer coefficient. To evaluate the cycle's cooling rate, the air volume flow rate from 400 to 3600 CMH is considered for coolant flow rate 10 to 30 liters per minute.



Figure 5. Cooling system Schematic diagram using radiator.


2.2.3 Heater Mode

An induction heater is employed in the heater mode to warm the coolant to extract the battery pack's heat. Figure 6 demonstrates the heating mode cycle to maintain the required narrow bandwidth in the cold climate region. The author of this study already studied the battery thermal management system's heating system in detail in Ref. [7,8].



Figure 6. Schematic diagram of the BTMS heating system using an induction heater



2.3 Chiller

The chiller is the cooling system's central component if the temperature is over 32 °C, which allows the coolant temperature to be further reduced by the refrigerant. A typical chiller diagram is shown in Figure 7. The coolant of a secondary circuit flows through the channels for this reason. The coolant has to be cooled in the chiller to the actual temperature following heat absorption. The temperature drop in the chiller occurs by the refrigerant's evaporation, which circulates in the chiller.



Figure 7. Simulation diagram of Chiller.



2.4 Radiator

A parallel flow heat exchanger is used as a radiator to transfer the heat to the environment. A radiator simulation diagram is shown in Figure 8.



Figure 8. Simulation diagram of a radiator.



2.5 Battery Pack

The nucleus of an integral component of the electric vehicle is the battery pack. A Power battery has drawn the bulk of researchers' interest as essential electric vehicle components. A 180 Ah prismatic lithium-ion pouch cell with 300 mm length, 98 mm width, and 14 mm thickness is considered for this study. The picture of the cell is shown in Figure 9. The battery pack consists of 5 modules with a total pack capacity of 64 kWh and a voltage of 356. The battery's position is on the vehicle's floor at the bottom, low the vehicle's center of gravity. The lower gravity significantly increases the vehicles' stability. Also spread across the base is the large battery pack, ensuring structural rigidity against side collisions.



Figure 9. Pouch cell diagram.

Battery length to width ratio is another important factor in the temperature distribution between the cells. A reason for non-uniform temperature distribution



between the cells is the ratio of battery length and width [45]. A suitable battery geometry design can produce a uniform distribution within cell-to-cell. A battery pack assembly from cell to vehicle integration is shown in Figure 10.



Figure 10. Battery pack assembly from cell to vehicle integration



2.6 Cooling System

A combination of water and ethylene glycol is considered as a coolant in this study. The reasons behind choosing the glycol with the water are: firstly, water will solidify if the temperature is lower than 0 °C, and secondly, severe battery damage may occur due to volume expansion of the frozen water. The thermophysical properties of coolants are itemized in Table 1.

Table 1. Operating fluid thermophysical properties

Thermal conductivity	0.3892 W.m ⁻¹ . K ⁻¹
Density	1069 kg.m ⁻³
Kinetic viscosity	2.58×10^{-6}
Specific heat capacity	3323 J.kg ⁻¹ . K ⁻¹

An aluminum-made cold plate is used underneath the battery to cool down the battery pack. The cooling passages in parallel formation are passing through the plate from all directions.

The diameter of the channel is also the primary parameter and should not be too small. If the channel diameter is too small, there will be a more massive pressure drop, and it will exert more load on the pump's energy. Cooling liquid flows into the channels. The distance between the two channels also impacts the uniformity of temperature



between the cells. Zhao et al. [42] studied large liquid-cooled lithium-ion battery packs and found that shorter cooling channels would significantly minimize temperature unevenness. Another parameter is the number of channels on a cold plate. If there are fewer channels, it will create a high-pressure drop and bad thermal efficiency. More channels cause higher costs. So, the number of channels should be optimal. Figure 11 displays a different kind of cold plate configuration for battery cooling.



Figure 11. Battery pack and coolant entrance and exit paths



The generated heat is transferred to the cold plate, an indirect method through a contact surface. A liquid heat-transfer medium is pumped through the channels. So, coolant comes and around, transferring the heat from the battery cells into the plate. The transferred heat to coolant then exchanges the heating with the chiller or radiator cycle depends on the coolant's initial required temperature. Besides, in terms of the heat transfer rate, a cold plate's cooling performance is determined.

An applicable device is required to energize the coolant. This research uses a small centrifugal pump order to pressurize the coolant. 5 and 30 LPM, respectively, were the lowest and highest flow rates. Meanwhile, the increasing flow rate of cooling liquid caused the pump's energy consumption to increase significantly because the pack itself primarily supplies the battery energy for the cooling system.





Figure 12. Different cold plate configurations



It is energy abuse of the pump when the temperature drop trend decreases with the flow rate increase. An optimal flow rate can be determined by considering the cooling efficiency and power consumption. The power usage by the pump versus the coolant flow rate is presented in Figure 12.



Figure 13. Energy consumption of pump versus flow rate.



Chapter 3 Simulation Setup

This chapter is about analysis setup details. A 45 kWh and 64 kWh rating batteries are considered to evaluate the indirect cooling performance for electric vehicles. To simulate the behavior of the proposed system, a standard simulation tool known as GT-SUITE is utilized. A chiller and radiator are being used to heat transfer between refrigerant and coolant and environment and coolant, respectively. A chiller with several power capacities 1, 2, and 3 kW is considered at the ambient temperature of 25, 30, and 35 °C. A combination of water and ethylene glycol is employed as a coolant in this study. The flow rate is considered between 5 LPM to 30 LPM. The initial battery temperature considered in this study was 30, 35, and 40 °C.

On the other hand, the airflow rate of 400 CMH to 3600 CMH is considered in the radiator cycle to remove heat to the environment. A coefficient of heat transfer evaluates the cooling rate. The heat transfer from fluids within the channels and flow split to their walls is determined using a heat transfer coefficient. The heat transfer coefficient is evaluated from the thermo-physical properties, fluid velocity, and the wall surface's roughness at any point in time. Using the Colburn analog, smooth pipes' heat transfer coefficient is measured using Equation (1).

$$h_g = \frac{1}{2} C_{f\rho} U_{eff} C_p P r^{\frac{-2}{3}}$$
(5)



Where C_f = fanning smooth pipe friction factor, ρ = density, U_{eff} = effective velocity of the outside boundary layer, C_p = specific heat, and Pr = prandtl number.

The Colburn analogy is considered for laminar, transitional, and turbulent flow. In advanced setup-miscellaneous, If the option improved laminar heat transfer model chosen, the heat transfer coefficient is determined from the constant Nusselt number in the laminar scheme. A Nusselt number of 3.66 is used for a Reynolds number less than 2000, and then the heat transfer coefficient is defined using Equation (2):

$$h_g = \frac{Nu.k}{d} \tag{6}$$

Where Nu = Nusselt number, k = thermal conductivity, and d = pipe diameter.

The value is interpolated between the constant Nusselt number at Re = 2000 and the value at Re = 4000 from the transformation regime's Colburn analogy. The Colburn analogy is utilized for the turbulent regime above Re = 4000.

The heat transfer rate versus coolant flow rate is investigated at different ambient temperatures and initial battery temperatures. On the other side, in the radiator cycle, the cooling rate is assessed as a flow rate function with varying airflow ratios at an ambient temperature of 25 °C. All the test conditions are listed in Tables 2 and 3, respectively.



Chiller Capacity (kW)	Flow Rate (LPM)	Ambient Temp (°C)	Initial Battery Temp (°C)
5 10 3 20 30			30
		25	35
	5		40
	10		30
	20	30	35
	30		40
		30	
		35	35
			40

Table 2. Test Conditions for Cycle using Chiller

Table 3. Test Conditions for Cycle using Radiator

Ambient Temp (°C)	Flow Rate (LPM)	Initial Battery Temp (°C)	Air Flow (CMH)
			1200
	10	30	
25	20	35	
	30	40	2400

3600



Chapter 4 Results and Discussion

4.1 45 kWh Battery Pack

4.1.1 Cooling Cycle using Chiller for 45 kWh Battery Pack



Figure 14. Cooling rate against flow rate at an outside temperature of 25 °C.

Figure 14 illustrates the cooling rate as a flow rate function at an ambient temperature of 25 °C and battery working temperatures of 30, 35, and 40 °C. The outside temperature is an essential parameter in battery thermal management. A chiller having a 3 kW of capacity is considered in this study. It depicts that the cooling rate increases from 5 to 20 LPM before decreasing at 30 LPM. At the initial stage, the



cooling rate rose dramatically, followed by 10 LPM. There is a slight increase in the cooling rate from 10 to 20 LPM. After that, the cooling rate dropped a little with the growth of flow rate to 30 LPM. It is clear to see that the maximum cooling rate is 0.0133 °C/sec at 40 °C battery temperature with a flow rate of 20 LPM. In comparison, the minimum is 0.0113 °C/sec at 30 °C of battery temperature with a flow rate of 5 LPM, which shows the impact of ambient temperature and flow rate on the system efficiency.

Figure 15 demonstrates the cooling rate concerning the coolant flow rate at the ambient temperature of 30 °C with the initial battery temperatures of 30, 35, and 40 °C. A chiller with 3 kW of power is employed in this study to cool down the coolant. The ambient temperature is one of the main factors involved in the battery thermal management system's performance. The maximum cooling rate is at initial battery temperature of 40 °C with the flow rate of 20 LPM is 0.0128 °C/sec, and the minimum is 0.0109 °C/sec at battery temperature of 30 °C with the flow rate of 5 LPM; displays the influence of ambient temperature and liters per minutes of flow rate on the overall performance. The overall cooling rate is decreased with the increase in ambient temperature. It reveals a significant increase in the cooling rate from 5 to 10 LPM. After that, the cooling rate slowly increases with the rise in coolant flow rate before a gradual decline followed by the maximum flow rate of 30 LPM.





Figure 15. Cooling rate against flow rate at an outside temperature of 30 °C.

Figure 16 presents the cooling rate's influence in terms of coolant flow rate at an ambient temperature of 35 °C with working battery temperatures of 30, 35, and 40 °C. The ambient temperature in this level is the maximum that has been used in this evaluation. The chiller power rating is 3 kW used to remove the heat from the coolant after heat absorption. The overall trend is almost the same as the previous ones but decreased the cooling rate because of increased ambient temperature. The maximum cooling rate found in this level is 0.0123 °C/sec with a flow rate of 20 LPM at 40 °C temperature of the battery pack, and the minimum is 0.0104 °C/sec at initial battery temperature of 30 °C with 5 LPM flow rate.



Although a given mass of coolant does have less time to absorb heat when coolant is being pumped in faster, now more coolant is being pumped in by the same factor. The difference is that this coolant mass does not get as hot when it leaves the plate, so it can absorb more heat from the plate per unit time as it flows.



Figure 16. Cooling rate against flow rate at an ambient temperature of 35 °C.







Figure 17. Cooling rate versus flow rate at an airflow of 1200 CMH.

Figure 17 presents the cooling rate impact on coolant flow rate at three different initial battery temperatures of 30, 35, and 40 °C. The airflow rate of 1200 cubic meters per hour is considered because it is a radiator cycle where forced air is employed to remove coolant heat to the ambient. The ambient temperature is set to 25 °C. The trend at all three-battery initial temperature is increasing. It is growing slowly to 20 LPM at 30 and 35 °C temperatures of battery pack then almost remained the same with little changes. On the other hand, at 40 °C, the cooling rate increasing rate is more than the different two temperatures, and the cooling rate is maximum at a flow rate of 30 LPM, which is 0.0146 °C/sec.



Figure 18 shows the cooling rate between coolant and battery pack against the flow rate of coolant. The considered initial battery temperatures are 30, 35, and 40 °C. Airflow rate of 2400 CMH is also considered because of the radiator using cycle. The coolant is get cooled by the ambient air in the radiator. The ambient temperature set for this level is 25 °C. The trends are showing an increase in cooling rate as a result of an increase in flow rate. With the increase in the coolant flow rate at the ambient temperature of 40 °C, there is a slight increase in the cooling rate. The highest rate in all three ambient temperature trends is 0.0184 °C/sec at 40 °C battery temperature and 30 LPM of flow rate, which shows the flow rate influence on the cooling rate.



Figure 18. Cooling rate versus flow rate at an airflow of 2400 CMH.



Figure 19 describes the effect of heat transfer rate against the coolant flow rate at three different initial battery pack temperatures of 30, 35, and 40 °C at an ambient temperature of 25 °C. An airflow rate of 3600 cubic meters per hour is considered because it is a cycle using a radiator. The cooling rate increases significantly at the ambient temperatures of 30 and 35 °C. While at the ambient temperature of 40 °C, the cooling rate rose dramatically to 20 LPM, there is a small rise followed by 30 LPM of flow rate. The maximum cooling rate is 0.0201 °C /sec at 30 LPM of flow rate and 40 °C of ambient temperature.



Figure 19. Cooling rate versus flow rate at an airflow of 3600 CMH.



4.2 64 kWh Battery Pack



4.2.1 Cooling Cycle using Chiller for 64 kWh Battery Pack

Figure 20. Impact of cooling rate with flow rate at ambient temperature 25 °C.

Figure 20 refers to the cooling rate as a function of operating fluid flow rate at an ambient temperature of 25 °C and the three different battery pack temperatures of 30, 35, and 40 °C. A chiller rating of 3 kW is employed in this analysis. The results display the fluctuations in the cooling rate at different conditions. For example, the cooling rate jumped to 10 LPM of flow rate at the initial stage. Then, there is a slight increase in the cooling rate before the decline with the same trend towards the flow rate of 30



LPM. The optimum cooling rate is at 40 °C with a flow rate of 20 LPM, which is 0.0959 °C/sec.

Figure 21 shows the systems' cooling rate with the coolant flow rate at an ambient temperature of 30 °C with three different initial battery temperatures of 30, 35, and 40 °C. A chiller having a capacity of 3 kW is employed to cool down the coolant temperature. The cooling rate increases between the 5 and 10 LPM of flow rate and decreases to 30 LPM at all different battery pack temperatures. The maximum cooling rate is 0.0092 °C/sec at the initial battery temperature of 40 °C with a 20 LPM flow rate.



Figure 21. Impact of cooling rate with flow rate at ambient temperature 30 °C.



Figure 22 refers to the cooling rate's impact with the coolant flow rate at three different initial battery pack temperatures of 30, 35, and 40 °C and an ambient temperature of 35 °C. A 3 kW of chiller power is used to remove the heat from the coolant. The trend of cooling rate is almost similar to the last one. The cooling rate rose significantly at the initial flow rate, and then there is a small increase in cooling rate per degree centigrade. After passing from 20 LPM, the cooling rate is decreased, followed by 30 LPM of flow rate.



Figure 22. Impact of cooling rate with flow rate at ambient temperature 35 °C.



4.2.2 Cooling Cycle using Radiator for 64 kWh Battery Pack

Figure 23 is about the cooling rate against the coolant flow rate at three initial battery pack temperatures of 30, 35, and 40 °C. The airflow rate of 1200 CMH is also considered in this analysis because of the radiator cycle. The ambient temperature was fixed to 25 °C. At 35 °C of temperature, the cooling rate is 0.0062 °C/sec at 10 LPM of flow rate, while at the 30 LPM, the cooling rate is 0.0077, which shows the increase in the with the increase in flow rate. The optimum point of cooling rate is at 40 °C with a flow rate of 30 LPM.



Figure 23. Cooling rate against flow rate at an airflow of 1200 CMH.



Figure 24 illustrates the system's cooling rate against the coolant flow rate at three different initial battery temperatures of 30, 35, 40 °C. The ambient temperature I set to 25 °C. The airflow rate is also considered because of the radiator cycle, which is 2400 cubic meters per hour. The trends show the increase in the cooling rate with the increase in the flow rate. As shown in the Figure, the cooling rate at 40 °C of the initial battery pack with a 20 LPM flow rate is 0.0143 °C/sec, and at 30 LPM, it is 0.0154 °C/sec, which shows the increase in the overall cooling rate.



Figure 24. Cooling rate against flow rate at an airflow of 2400 CMH.





Figure 25. Cooling rate against flow rate at an airflow of 3600 CMH.

Figure 25 demonstrates the heat transfer rate versus the coolant flow rate at three different initial battery pack temperatures. The initial battery temperatures considered for this analysis are 30, 35, and 40 °C. The ambient temperature was also fixed to 25 °C. This analysis is based on the cycle using a radiator to cool down the coolant temperature, so an airflow rate of 3600 CMH is employed. The increasing trend rate is different for battery pack temperatures. Anyhow, the cooling rate increases with the increase in the coolant flow rate. At the 30 and 35 °C temperatures, the increasing trend is almost similar, while at the 40 °C, the cooling rate is increasing dramatically. The cooling rate's optimum value is at 40 °C with a flow rate of 30 LPM, which is 0.0172 °C/sec.



Chapter 5 Comparison and Verification

In this chapter, the comparison of 45 kWh and 64 kWh battery packs is made to evaluate an indirect cooling system's performance. Firstly, a comparison made according to a cycle using a chiller to cool down the coolant. The results of both battery packs with the test conditions are demonstrated in Figure 26. For the chiller cycle, results from a chiller capacity of 3 kW at 40 °C initial battery temperature with a flow rate of 20 LPM condition is utilized for battery packs comparison.







In the radiator cycle, results from a coolant flow rate of 30 LPM at 25 °C of ambient temperature and 40 °C of initial battery temperature are employed to compare two battery packs, 45 & 64 kWh, respectively. Thus, as shown in the Figure, a 45 kWh rating battery pack maintains a higher cooling rate than 64 kWh. A 45 kWh battery has a cooling rate of 0.01336 °C/sec at 25 °C of ambient temperature, 40 °C of initial battery pack temperature, and 20 LPM of flow rate with a 3 kW chiller capacity. While with the same parameters, the 64 kWh battery pack's cooling rate is 0.00959, which shows the 45 kWh battery pack's efficiency. It is because of the heat generated inside the pack. If there is a need to increase the cooling rate and efficiency of a 64 kWh battery pack, a high rated chiller or more LPM of flow rate is required.

The Figure also compares battery packs for a radiator cycle to determine the optimum values. In this cycle, a radiator is used to remove the heat from the coolant. The cooling rate of a 45 kWh battery pack at an ambient temperature of 25 °C and initial battery temperature of 40 °C with a coolant flow rate of 30 and airflow rate of 3600 CMH is 0.0201 °C/sec. While following the same conditions, this value is 0.0172 °C/sec for the 64 kWh battery pack. The result shows that a 45 kWh battery pack is dominant in terms of performance. To sum up, it is evident that a 45 kWh battery pack is efficient compared to a 64 kWh battery pack in both chiller and radiator cycles.

In one study, Shen & Goa [46] performed a system simulation on refrigerant-based battery thermal management technology for electric vehicles. They used a cold plate to remove the heat from the battery. The initial battery temperature assumption were 31.5, 36.5, & 41.5 °C and found the cooling rate of 0.0121, 0.0191, 0.0263 °C/sec,



respectively. On the other hand, in the proposed study, the initial battery temperature is 30, 35, & 40 °C, and the obtained results are 0.0123, 0.0128, and 0.0133 °C/sec, respectively, which show the accuracy of the results.

Figure 27 displays the vehicle's real experimental model to assess the proposed cooling system's performance, similar to the experiment results.





Figure 27. Experimental diagram for simulation results verification



Chapter 6 Conclusions and Future Work

To maintain the optimal temperature range and the local temperature difference between the cells, an indirect cooling system based on cold plates is proposed in this study. Two different rated battery pack, 45 kWh, and 64 kWh, are considered in this study. The results and discussion about the study are discussed in chapter 3. Herein, to conclude the investigation, this chapter is made to know about the proposed research.

In the chiller cycle, the cooling rate increases with an increase in the fluid flow in but to some extent, after a certain point, there is a decrease in the cooling rate because of coolant pressure and less time to absorb heat. Also, chiller capacity has a substantial effect on the cooling system. In the radiator cycle, the cooling rate increases throughout. There is no decrease in the cooling rate because air is used to cool down the coolant temperature with the various flowrate in this cycle air.

In comparing 45 & 64 kWh battery packs, a 45 kWh rated battery pack is found batter in heat transfer efficiency in both chiller and radiator cycles. In the chiller cycle, a 45 kWh battery pack has a maximum cooling rate of 0.01336 °C/sec at an ambient temperature of 25 °C and an initial battery pack temperature of 40 °C with a coolant flow rate of 20 LPM and 3 kW chiller capacity. On the other hand, a 64 kWh battery back has a maximum cooling rate of 0.00959 °C/sec at the same conditions. In the radiator cycle, a 45 kWh battery pack is again dominant in terms of heat transfer rate.



The maximum cooling rate is 0.02012 °C/sec at 25 °C ambient temperature and 40 °C initial battery pack temperature with a coolant flow rate of 30 LPM and an airflow rate of 3600 cubic meters per hour. For a 64 kWh battery pack, the maximum heat transfer rate is 0.01726 °C/sec with the same working conditions. Hence, the numbers proved that a 45 kWh rated battery pack is efficient in both cycles for the battery thermal management system. Finally, a BTMS with 3 kW of chiller capacity and a coolant flow rate of 20 LPM at an ambient temperature of 25°C and 40°C temperature of the battery pack is suggested for the chiller-based cycle. For the radiator-based cycle, a BTMS with an airflow rate of 3600 cubic meters per hour and 30 LPM of coolant flow rate at an ambient temperature of 25°C and 40°C temperature of the battery pack is suggested.

The cooling system has many benefits such as maximum temperature drops, temperature uniformity between the cells, top performance, applicable to all kinds of cells, reduction in noise level, and highly commercialized in the current electric vehicles. In contrast, there are also some drawbacks to the cooling system. For example, the proposed cooling system can potentially leakage, increase the battery pack's weight, high cost, and maintenance.

In terms of benefits, drawbacks, and efficiency of the proposed system, it is concluded that a hybrid system build as a couple of two separate thermal management systems will be a stable, modular system for battery pack thermal management. Another approach to increase the cooling rate is suspending the Nano-sized solid particles (Al,



Cu, Ni, Ag, AL₂O₃, TiO₂, & Fe₃O₄) to increase the thermal conductivity of the coolant, which is beneficial to reduce energy consumption.



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